

Band alignment at InP/TiO₂ interfaces from density-functional theory

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Abstract

The natural band alignments between indium phosphide and the main dioxides of titanium, i.e. rutile, anatase, and brookite as well as amorphous titania are calculated from the branch-point energies of the respective materials. Irrespective of the titania polymorph considered, type-I band alignment is predicted. This may change, however, in dependence on the microscopic interface structure: supercell calculations for amorphous titania grown on P-rich InP(001) surfaces result in a titania conduction band that nearly aligns with that of InP. Depending on the interface specifics, both type-I band and type-II band alignments are observed in the simulations. This agrees with recent experimental findings.

Keywords: interfaces, density functional theory, band alignment, InP, TiO₂, molecular dynamics

1. Introduction

Indium phosphide is frequently used in optoelectronics and high-speed electronics, see, e.g. [1, 2]. The InP band gap is close to the optimum for solar energy conversion [3]. Therefore, the material is also increasingly used for photovoltaics [4] as well as in photoelectrochemical devices for water splitting [5]. In such devices, a thin TiO₂ film is often used as a corrosion protection layer [5–8]. The InP/TiO₂ heterojunction has also emerged as a promising candidate for solar cells [9] as well as for photocatalytic applications [10, 11].

While the structural and electronic properties of the InP growth plane are well understood [12–15], comparatively little is known about the InP/TiO₂ interface. This holds even for fundamental properties such as the interface band alignment.

The band alignment, however, is crucial in determining how efficiently charge carriers (electrons and holes) move across the interface between the two materials. It will affect not only charge carrier flow and energy barriers, but also device efficiency and interface stability. Experimentally, it was found that TiO₂ serves as an effective hole blocking layer on the surface of InP while allowing the transport of electrons [16]. This would be consistent with a type-II junction. Yin *et al* [9], on the other hand, found the TiO₂ valence-band maximum (VBM) and conduction-band minimum (CBM) to be 2 eV below and 0.05 eV above the respective value of InP, corresponding to a type-I junction. Recently, Cipriano *et al* [17] used density-functional theory (DFT) to study microscopically the interface between the InP cleavage plane (110) with thin anatase TiO₂ films. Irrespective of the film thickness and irrespective of the TiO₂ surface orientation, a type-II band alignment was calculated, with the anatase CBM 0.10...0.64 eV below that of InP. This agrees with the experimental findings of [16], but does not necessarily contradict the measurements of Yin *et al* [9]:

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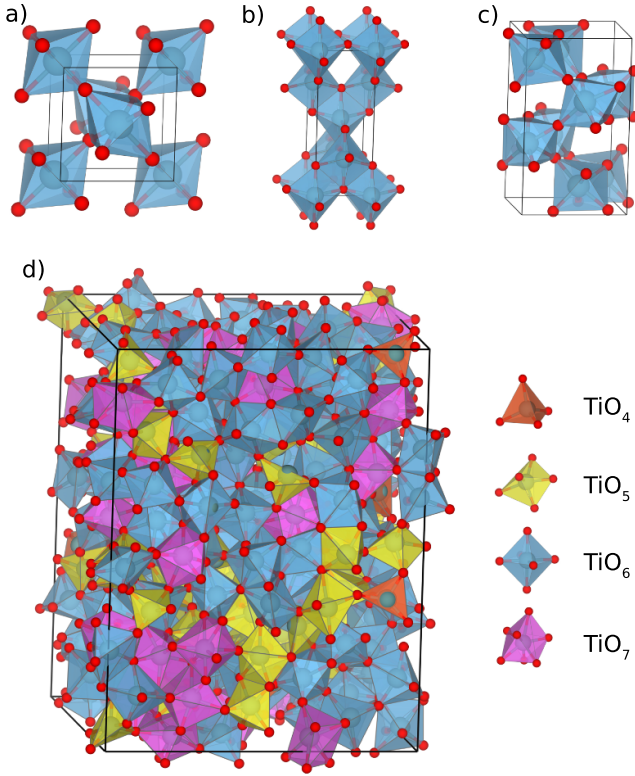


Figure 1. Polyhedra representation of the TiO_x building blocks used to model the (a) rutile, (b) anatase, and (c) brookite polymorphs of titania. An example structure of a- TiO_2 is shown in (d).

Anatase and rutile titania form a type-II interface, where the rutile CBM is 0.3...0.5 eV above the corresponding value of anatase [18–20]. As Yin *et al* employed atomic layer deposition (ALD) to deposit titania on InP, their measurements cannot directly be compared with calculations for anatase.

In order to better understand and to optimize the InP/ TiO_2 interface, knowledge of the band alignment between the main dioxides of titanium, i.e. rutile, anatase, and brookite as well as amorphous titania (a- TiO_2) with indium phosphide is required. Corresponding data, based on DFT, are provided in the present work. On the one hand, the branch point energy (BPE) or charge neutrality level method [21–23] was used. The BPE method requires no knowledge of the structural details of the interface and provides accurate natural band alignments. On the other hand, the technologically particularly relevant case of an a- TiO_2 film deposited on P-rich InP(001):H—the InP surface typically obtained from gas phase epitaxy—is studied with microscopic supercell calculations.

2. Methodology

In the present work, the common crystalline titanium dioxide polymorphs rutile, anatase and brookite along with amorphous TiO_2 are studied, see figure 1. The atomic structures of the crystalline polymorphs are determined using DFT within the Vienna *ab-initio* Simulation Package (VASP) [24] implementation. Electron-ion interactions are described using the

projector-augmented wave approach [25, 26]. In addition to the 3d and 4s states, also the 3s and 3p electrons are treated as valence states for Ti. Similarly, the In 4d states are considered in the valence shell, in addition to 5s and 5p. The PBE functional [27] within the general gradient approximation is used to calculate the exchange and correlation energy.

In order to generate atomic structures that model amorphous TiO_2 , we start with classical molecular dynamics (MD) using the force field of Matsui and Akaogi [28] applied to a $4 \times 4 \times 8$ crystalline rutile supercell at 10 000 K. Subsequent to an 800 ps classical MD simulation run with the LAMMPS code [29], the structure was subjected to 10 ps *ab initio* MDs (AIMD) on the DFT level of theory using CP2K [30]. The AIMD was done using DZVP-MOLOPT-SR-GTH basis sets, Goedecker–Teter–Hutter pseudopotentials [31, 32] and the PBE functional [27] together with the empirical dispersion correction scheme (D3) of Grimme [33]. The temperature is set to 350 K applying a Nose–Hoover chain thermostat [34–36]. In order to obtain a representative ensemble of possible amorphous conformations, 10 equidistant snapshots were selected from the AIMD trajectory for detailed analysis of the electronic properties. The resulting amorphous bulk structures consist of 768 atoms in a tetragonal cell with lattice constants $a = 18.4$ and $c = 23.67$ Å. They are characterized by mostly 6-coordinated Ti ions with lesser amounts of 5 and 7-coordination and even fewer 4-coordinations. The average coordination number for Ti is 5.86, close to previous simulations [37, 38].

The InP/a- TiO_2 interface was modeled starting from the energetically favored InP(001)(2×2)2D–2H surface, the equilibrium surface structure resulting from gas-phase epitaxy [14, 15]. The InP slab consists of 12 layers. The bottom is passivated with partially charged pseudohydrogen ($Z = 1.25$). Computationally, the calculations are on the same footing as used in our previous InP(001) surface calculation, and well converged with respect to the numerical parameters [15, 39, 40]. A lateral (4×4) periodicity was chosen to allow for a realistic interface with a- TiO_2 . The structure of the ~ 18 Å thick amorphous titania overlayer containing 444 atoms was again obtained from MDs. Here, a protocol similar to the one described above for the preparation of bulk a- TiO_2 was used: Starting from the InP slab, additional TiO_2 building blocks were added to the simulation box, corresponding to the rutile structure. They were first subject to a classical MD at 2000 K—with the InP structure frozen - and subsequently thermalized using AIMD, as described above.

Natural band offsets of bulk materials can be obtained by aligning the branch point energies of the respective systems. Here the BPE are calculated following the formalism proposed, *inter alia*, by Schleife *et al* [41]. This approach has already successfully been applied to TiO_2 [18]. In detail, the BPE is obtained from

$$E_{\text{BP}} \approx \frac{1}{2N_k} \sum_k \left[\frac{1}{N_{\text{VB}}} \sum_i^{N_{\text{VB}}} \varepsilon_{v_i} + \frac{1}{N_{\text{CB}}} \sum_j^{N_{\text{CB}}} \varepsilon_{c_j} \right], \quad (1)$$

Table 1. Band gap energies calculated here in comparison to measured data.

Material		Band gap (eV)		
		PBE	HSE06	Exp.
InP		0.44	1.40	1.34 ^a
TiO ₂	Rutile	1.82	3.32	3.3 ^b , 3.6 ^c
	Anatase	2.09	3.59	3.8 ^d
	Brookite	2.33	3.82	3.4 ^e
	a-TiO ₂	1.58	3.07 ^f	

^a Tian *et al* [46].^b Tezuka *et al* [47].^c Rangan *et al* [48].^d Wang *et al* [49].^e Koelsch *et al* [50].^f Extrapolated using the shift obtained for crystalline TiO₂.

where N_k is the total number of points used for Brillouin zone sampling and ε_{v_i} and ε_{c_j} are i th highest valence and j th lowest conduction band energies, respectively. The number of valence (N_{VB}) and conduction states (N_{CB}) that need to be included in equation (1) for converged results depends on the details of the band structure. For III–V compounds about two valence states and about one conduction state per two atoms (one of each species) is typically sufficient [41, 42]. Deak and co-workers used up to 12 valence and 10 conduction states per two TiO₂ units [18]. This approach has been followed here. Hybrid DFT using the HSE06 functional [43] as implemented in VASP [24] is used for an accurate description of the electronic structure, and in particular the band gap. The a-TiO₂ calculations involve averaging over AIMD snapshots obtained within a period between 0.25–4 ps. For the InP/a-TiO₂ interface the average was taken for a substantially longer period between 0.5–17 ps.

Deep core states were used in order to obtain the band offsets in the microscopic calculations for the InP(001)/a-TiO₂ interface. In detail, we used the 3s and 4d energies of Ti and In, respectively, to align the VBM in the respective bulk and supercell structures. Alternatively, a microscopic analysis can be performed based on the spatial projection of the electronic band structures, see, e.g. [44, 45].

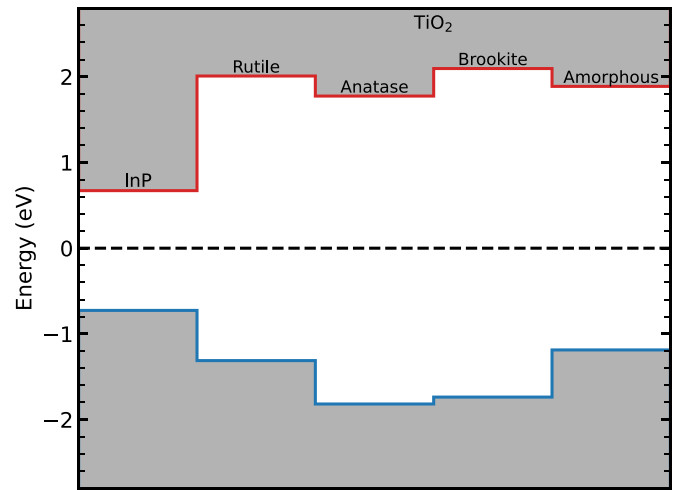
3. Results and discussion

The fundamental bulk band gaps calculated here for the various materials are shown in table 1. The hybrid DFT results for InP and rutile as well as anatase compare well with experiment [46–49]. The apparently slightly larger deviations between experiment and theory for brookite may be related to the difficulties encountered to obtain it as a pure phase. In addition, it should be mentioned that the values derived in [49, 50] are based on optical measurements and will therefore be affected by excitonic effects [51].

The hybrid DFT band gap values in table 1 together with the hybrid BPE calculated from equation (1) are used to derive the natural band offsets. In case of a-TiO₂, the 768 atom supercell

Table 2. Branch point energies and natural band offsets vs InP bulk for various TiO₂ phases calculated within hybrid DFT. All values in eV.

Material		BPE	ΔE_{VB}	ΔE_{CB}
InP		0.73		
TiO ₂	Rutile	1.31	−0.59	1.34
	Anatase	1.82	−1.09	1.10
	Brookite	1.73	−1.00	1.43
	a-TiO ₂ ^a	1.19	−0.46	1.22

^a Values extrapolated using the shifts obtained for crystalline TiO₂.**Figure 2.** Natural band alignment between InP and various TiO₂ polymorphs calculated within hybrid DFT. The a-TiO₂ values are extrapolated.

does for numerical reasons not allow for hybrid DFT calculations. However, given that both the band gap openings as well as the shifts of the BPE upon replacing PBE by HSE06 are essentially equal for all crystalline titania phases, we apply the same shifts to a-TiO₂. The natural band offsets thus calculated are summarized in table 2. Here, the valence band (ΔE_{VB}) and conduction band offsets (ΔE_{CB}) are referenced to InP bulk. The BPE values are consistent with previous studies for TiO₂ [18] and InP [52].

The data from table 2 are visualized in figure 2, where the CBM and VBM are shown, all values are referenced to the BPE. The band gaps calculated for the various titania bulk phases considered here scatter within 0.7 eV. The scatter of the branch point energies is similar, around 0.6 eV. Nevertheless, all natural band offsets calculated are consistent with the type-I band alignment. This agrees with the data measured by Yin *et al* [9].

The natural band offsets calculated here are guidelines only for the actual band offsets, which will be affected by interface dipoles, interface strain as well as chemical bonding and stoichiometry changes at real interfaces. In order to see to what extent these effects will influence the interface band alignment, we perform microscopic calculations using DFT-PBE for one particularly relevant system, the interface between the

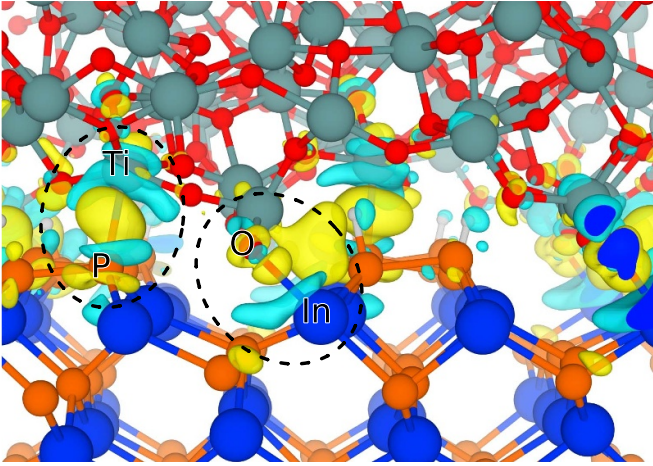


Figure 3. Atomic structure of the InP(001)/a-TiO₂ interface illustrating an example snapshot obtained from MD. Also shown is the charge density difference with respect to the separate bulk systems. Charge accumulation (depletion) is depicted in yellow (cyan). Examples of Ti–P and In–O bond formations are indicated.

P-rich InP(001) surface, i.e. the growth plane typically prepared using vapor phase epitaxy [40, 53, 54], and amorphous titania, as typically obtained from ALD [8, 55, 56]. For computational reasons, we cannot employ hybrid DFT to account for the quasiparticle energy corrections in all AIMD snapshot configurations. Rather, we use the extrapolation scheme applied earlier to a-TiO₂ bulk as presented in table 2. In order to evaluate the validity of this approximation, hybrid DFT calculations were performed for a single AIMD snapshot configuration. It was found that the extrapolation scheme agrees within an error bar of less than 0.2 eV with the hybrid DFT results.

An example AIMD snapshot of the interface is shown in figure 3. It can be seen that TiO₂ units occupy the trenches between the P dimer rows. The formation of In–O bonds with lengths in the range of 2.2–2.5 Å is observed from the calculated charge redistribution. Still, the essential features of the InP(001) surface remain intact. This agrees with the interpretation of recent x-ray and ultraviolet photoelectron spectroscopy data measured on titania films grown on InP(001) via ALD [8]. Cipriano *et al* [17] studied computationally the interface between InP and crystalline (anatase) titania. They observed the formation of In–O bonds and Ti–P bonds with lengths of about 2.3 and 2.7–2.8 Å, respectively. This is confirmed by the present simulations, where in addition to In–O bonds also Ti–P bonds are formed, see figure 3.

The AIMD interface snapshot configurations are used to calculate the interface band offsets microscopically. This is done by aligning the energies of the 3s and 4d energies of Ti and In in bulk structures with the respective values in the interface supercell. The average results are shown in figure 4. It can be seen that compared to the natural band alignment a substantial down shift of the titania band edges with respect to InP occurs. It amounts to 1.14 eV. Still, the calculated band

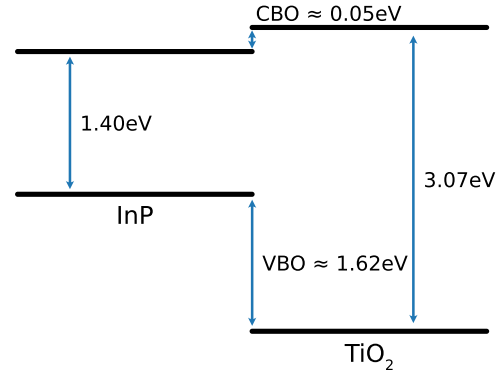


Figure 4. Band alignment of the InP(001)/a-TiO₂ interface calculated from the average of AIMD simulation snapshots. The In 4d and Ti 3s states were used to align the VBM of the InP and TiO₂, respectively.

alignment remains type-I. However, the conduction band offset is extremely small and in some AIMD snapshot configurations the alignment is type-II. This agrees with our calculations for the valence and conduction band offsets for the various realizations of a-TiO₂ with respect to InP bulk: The results obtained for the snapshot configurations are characterized by standard deviations of 0.28 and 0.23 eV for ΔE_{VB} and ΔE_{CB} , respectively. This is well above the conduction band offset determined here microscopically for the InP(001)/a-TiO₂ interface. Obviously, the specific realization of the amorphous TiO₂ film in dependence on the actual preparation conditions [37, 38] will noticeably affect the band alignment. Both, type-I and type-II heterojunctions may occur for InP/a-TiO₂ interfaces. The present computational results are in accord with the experimental observations, where depending on the InP(001)/a-TiO₂ interface preparation type-II [8, 16] as well as type-I alignment [9] is found.

However, not only the properties of the amorphous titania film will affect the band alignment. It will also depend on the interface dipole [57]. Given that the InP(001) surface dipole varies by as much as one eV in dependence on the surface stoichiometry [39, 58], indeed a very wide range of band offsets can be realized.

4. Summary and conclusions

The present calculations predict natural band offsets corresponding to a type-I band alignment for InP/TiO₂ interfaces, irrespective of the titania polymorph. However, there is an appreciable scatter of the band offsets. The valence band offset is largest for anatase, −1.1 eV, and smallest for a-TiO₂, −0.5 eV. The conduction band offset varies between 1.1 eV for anatase and 1.4 eV for brookite. Even larger changes can be expected to result from the microscopic structure of the interface. In case of the technologically relevant interface between a-TiO₂ and P-rich InP(001), the calculations predict a valence band offset of −1.6 eV, while the conduction band offset is essentially quenched. Depending on the properties of

the amorphous titania film, both type-I and type-II interfaces can be realized. The results presented here thus demonstrate that by controlling the microscopic interface structure and the atomic order of the titania film a very wide range of band alignments becomes accessible.

Data availability statement

All data that support the findings of this study are included within the article (and any supplementary files).

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Conflict of interest

The authors declare no conflict of interest.

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